

EVALUATION OF ANODIC Ta₂O₅ AS THE DIELECTRIC LAYER FOR EWOD DEVICES

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ABSTRACT

We report that anodic tantalum pentoxide (Ta₂O₅) exhibits severe polarity and frequency dependencies that, when used as the dielectric material for electrowetting on dielectric (EWOD) devices, in many cases result in long-term performance that is worse than when conventional dielectrics (e.g., SiO₂) are used. This rather disappointing news is nevertheless relevant to the community and calls for critical assessment. Here, we find that under direct current (DC) actuation Ta₂O₅ is attractive for EWOD only if the droplet is negatively biased, and that under alternating current (AC) it is acceptable only for low frequencies.

INTRODUCTION

A liquid wets or spreads on a dielectric surface when an electric field is applied between the liquid and the electrode underneath the dielectric layer. Known as electrowetting-on-dielectric (EWOD), this actuation mechanism has been the basis for a number of micro devices in various application areas: lab-on-a-chip [1-2], optics/display [3-4], electric switch [5] and rheometry [6]. For EWOD to be widely used in commercial products, high actuation voltages need to be avoided. The operation voltage can be lowered by using a dielectric with higher dielectric constant and/or reducing its thickness, as evident in the Lippmann-Young Equation below and experimentally confirmed [7].

$$\cos \theta_v - \cos \theta_0 = \frac{\epsilon_0 \epsilon_r V^2}{2\gamma t_r} \quad (1)$$

where θ_v and θ_0 are the contact angles with and without voltage, respectively, ϵ_0 is the permittivity of vacuum (8.85×10^{-12} F/m), ϵ_r and t_r are the dielectric constant and thickness of the dielectric layer, respectively, γ is the surface tension at the liquid-fluid interface (the fluid is air in this paper), and V is the voltage across the dielectric layer. High dielectric constant material is a better choice than reduced dielectric thickness because the latter brings with it the drawback of reduced breakdown voltage.

Tantalum pentoxide, as a relatively high dielectric constant (k) material (i.e., 18.5-27.5 [8, 9]), can reduce overall electrowetting voltage compared to common dielectrics (SiO₂ and Parylene around 2-3). Anodic Ta₂O₅, widely used in electrolytic capacitors with reproducible thickness and good film quality [10], has recently been reported as an alternative dielectric for its room-temperature fabrication, high- k , and high film quality (i.e., pinhole-free) [11]. However, the above study considered only DC actuation with a negatively biased droplet. In this report, we investigate the reliability of anodic Ta₂O₅ as an EWOD dielectric by testing additional actuation conditions. As a reference dielectric, we also tested thermal SiO₂, chosen for its stability across varied

fabrication conditions [7].

DEVICES AND EXPERIMENT

In order to evaluate Ta-Ta₂O₅ EWOD devices, Si-SiO₂ EWOD devices were used as control samples for comparison. Figure 1 shows the two types of EWOD devices used in the long-term testing experiments: (a) Ta-Ta₂O₅ and (b) Si-SiO₂ EWOD devices. In order to ensure dielectric film quality, anodic Ta₂O₅ and thermal SiO₂ were chosen.

Long-term performance of an EWOD system was determined by monitoring contact angle change ($\Delta\theta$) of a droplet between before and after voltage application throughout repeated electrowetting actuation. Slower reduction of $\Delta\theta$ implies better reliability of the EWOD device.

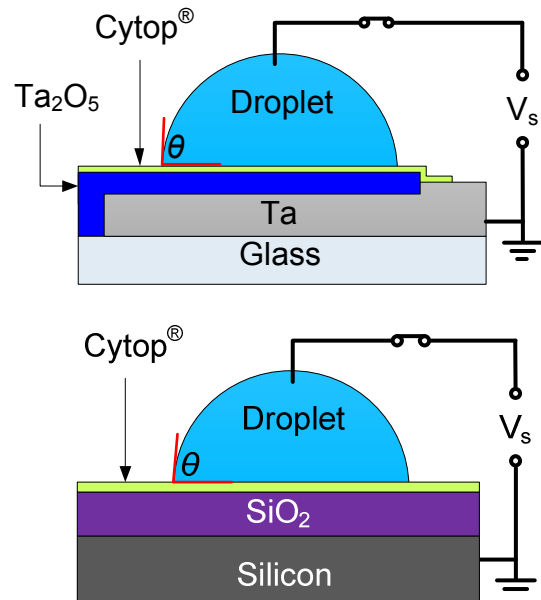


Figure 1: The Ta-Ta₂O₅ (top) and Si-SiO₂ (bottom) EWOD devices tested for repeated EWOD actuations

Fabrication

Process flows for fabricating Ta-Ta₂O₅ and Si-SiO₂ EWOD devices are shown in Figure 2 and Figure 3, respectively. Tantalum of 375 nm thickness was sputtered on clean 2.5 cm × 2.5 cm square glass microscope slides. A 200 nm-thick Ta₂O₅ film was grown on the tantalum by anodizing in 0.01 wt% citric acid bath using platinum as the counter electrode at room temperature [10]. During anodization, constant current density of 0.8 mA/cm² was applied until the voltage reached 100 V, after which 100 V was held for 30 minutes. Constant-current anodization was used to ensure a fixed value of anodization constant (20 Å/V), and the constant voltage phase was used to ensure final desired film thickness [12]. The film thickness was measured by NanoSpec (AFT 010-0180)

using refractive index of 2.22 reported in [8].

Separately, 560 nm-thick SiO₂ film was thermally grown on a p⁺⁺ Si wafer (with resistivity < 0.0015 Ω-cm) by wet oxidation at 1050°C for 56 minutes. Front side oxide was protected by photoresist AZ5214 during removal of the backside oxide in buffered oxide etching (BOE) solution. The photoresist was then removed in ALEG-380 stripper.

A 1.3% Cytop[®] solution was made by diluting 9% Cytop[®] solution with Cytop[®] solvent (1:6). The resulting 1.3% Cytop[®] solution was spin-coated on both dielectric layers at 500 rpm for 5 s followed by 1500 rpm for 30 s, giving a 50 nm-thick Cytop[®] film. After spin-coating, the film was baked at 150°C for 10 minutes to remove the solvent and 195°C for 1 hour to anneal.

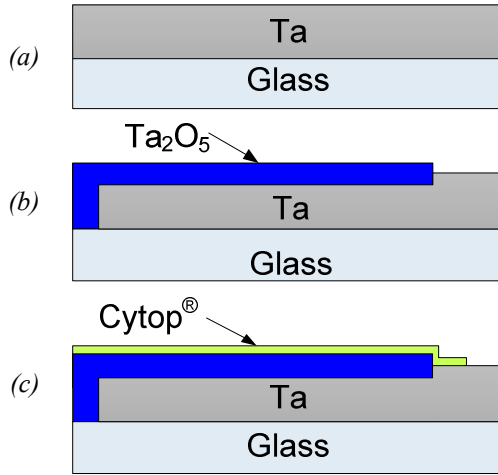


Figure 2: Process flow for Ta-Ta₂O₅ EWOD device. (a) 375 nm of Ta film was sputtered on a glass slide. (b) 200 nm of Ta₂O₅ was anodically grown in 0.01 wt% citric acid bath at 100 V. (c) A 50 nm Cytop[®] film was spin-coated.

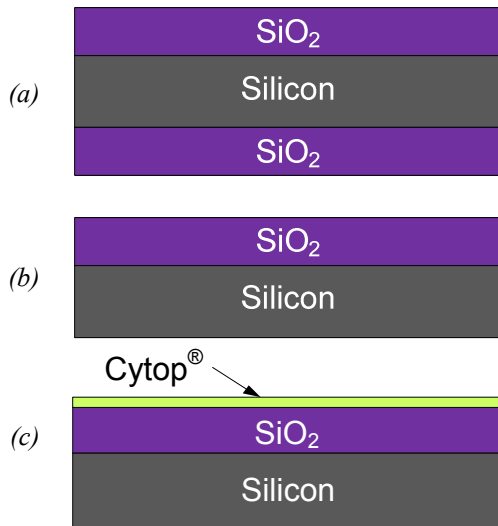


Figure 3: Process flow for Si-SiO₂ EWOD device. (a) 560 nm of SiO₂ film was thermally grown on a p⁺⁺ Si wafer. (b) Backside oxide was removed in BOE. (c) A 50 nm of Cytop[®] film was spin-coated.

EWOD Experiments

Contact angle measurement of a droplet during EWOD actuation was observed through a CCD camera (PixeLink, model PL B742U). Actuation voltage signals were generated by a LabVIEW program, outputting to a data acquisition (DAQ) system (National Instruments) and amplified to desired voltage values. Voltage was on for 3 seconds and off for 3 seconds, repeating for up to 2000 cycles. Figure 4 shows two images of droplet contact angle captured for each cycle: (a) without and (b) with voltage.

All experiments were conducted using a 10 μL droplet with electrical conductivity of 0.013 S/m. The droplet consisted of 50%v/v glycerin and 50%v/v KCl standard solution (Fluka, σ = 0.1413 S/m). The droplet was conductive enough to ensure the circuit's voltage drop occurred predominantly across the hydrophobic and dielectric layers, not across the droplet.

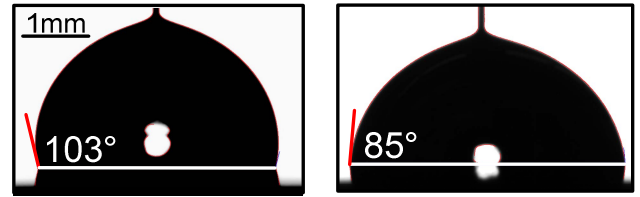


Figure 4: Images of contact angle measurement captured by CCD camera and analyzed by a contact angle measurement program for cases without voltage (left) and with voltage (right).

Three different types of actuation were used for long-term electrowetting testing: +DC, -DC, and an AC square wave, as shown in Figure 5(a-c). The frequency range used for AC actuation was from 50 Hz to 1kHz. In this paper, +DC means positive potential was applied to the droplet through a platinum wire while grounding the buried electrode (e.g., tantalum or silicon), as shown in Figure 5(a). Conversely, applying negative potential to the droplet is denoted as the -DC case, illustrated in Figure 5(b). A square wave was selected over other AC waveforms since its peak value is equal to its root-mean-square (RMS) value, which was set equal to the absolute DC voltage value used in the DC voltage experiments (i.e., $V_{pk} = V_{RMS} = |V_{DC}|$). Hence, maximum force acting on the triple contact line of solid-liquid-air is the same for both DC and AC cases.

In order to start with the same contact angle reduction ($\Delta\theta$), 11.6 V was applied to the Ta₂O₅ and 30 V to the SiO₂ device, corresponding to the same electrowetting number (i.e. $E_w = 0.31$). Electrowetting number represents the degree of EWOD actuation, as shown in the modified Lippmann-Young Equation below.

$$E_w = \cos \theta_v - \cos \theta_0 = \frac{\epsilon_0 \epsilon_r \epsilon_{Cytop}}{2\gamma(\epsilon_r t_{Cytop} + \epsilon_{Cytop} t_r)} V^2 \quad (2)$$

Here ϵ_{Cytop} and t_{Cytop} are the dielectric constant and thickness of the Cytop[®] layer, respectively, and V is the voltage across the dielectric and Cytop[®] layers together.

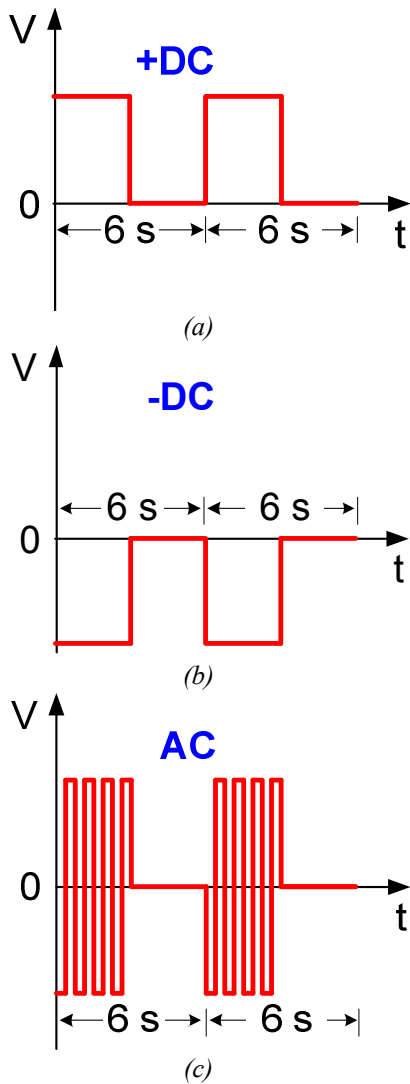


Figure 5: Actuation signals for long-term electrowetting testing with (a) +DC, (b) -DC, and (c) AC square wave.

RESULTS AND DISCUSSION

The principal piece of information to report about the Ta_2O_5 EWOD device is its utterly poor performance when the droplet is positively biased (i.e., +DC case), an important ramification not described in [10]. Electrolysis occurred immediately once +DC was applied, ruining the device. This dramatic polarity effect on Ta_2O_5 EWOD devices may be understood from the well-known diode-like behavior of Ta- Ta_2O_5 capacitors. Directly underneath the Ta_2O_5 outer surface, there exist negative charges induced by the anodization process [13], providing greater resistance to additional incoming electrons in the -DC case than the +DC case. In other words, current can easily flow from the droplet through the Cytop[®] and Ta_2O_5 layer to the tantalum electrode when the droplet is under positive potential. The sudden increase of current in the dielectric causes dielectric breakdown [12], visually observed as electrolysis.

Figure 6(a) shows the contact angle with and without DC voltage for three cases (-DC with Ta_2O_5 , +/- DC with SiO_2) up to 2000 cycles or 200 minutes of EWOD actuation. In the case of -DC, the on-state angles of Ta_2O_5

(86-88°) are almost the same as those of SiO_2 (86-88°), while the off-state angles of Ta_2O_5 (98-102°) are mostly lower than those of SiO_2 (100-104°). Figure 6(b) presents the same results expressed as the contact angle reduction ($\Delta\theta$) from the off to the on state, i.e., the effectiveness of electrowetting. The contact angle change on the Ta- Ta_2O_5 device decreases faster than that on the Si- SiO_2 device, implying that Ta_2O_5 degraded faster than SiO_2 under repeated EWOD actuations. Although the electrowetting effect diminishes faster with +DC than with -DC on most EWOD devices including the SiO_2 device [14], the polarity dependence is extreme for the Ta_2O_5 devices. While reasonable performance is seen on the Ta_2O_5 device with -DC throughout 2000 cycles, no results were attainable under +DC due to electrolysis.

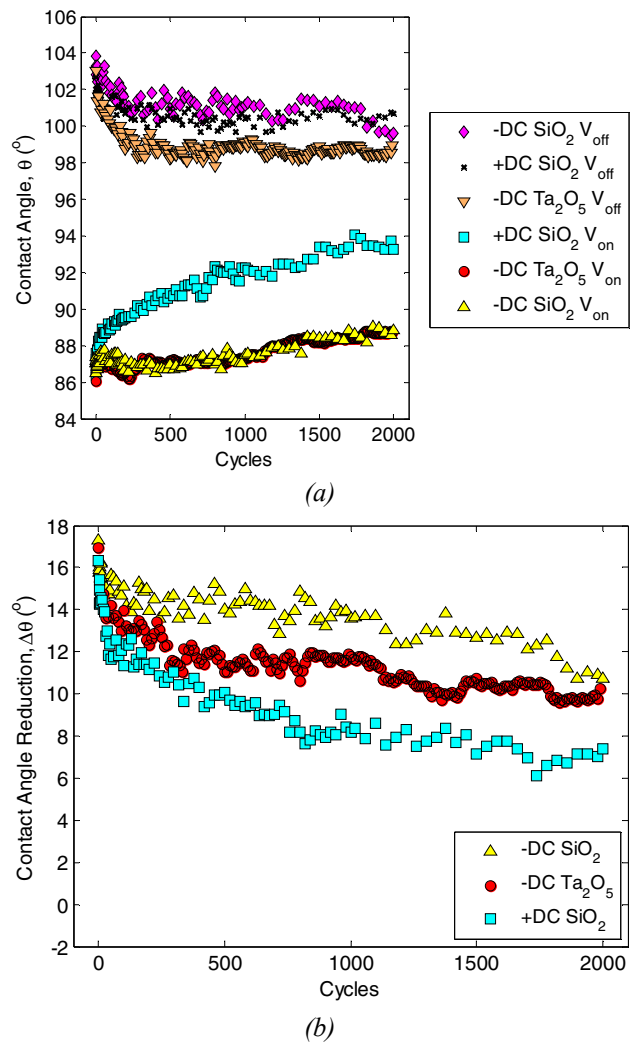


Figure 6: Measured data of (a) contact angle and (b) electrowetting-induced contact angle reduction on Ta_2O_5 and SiO_2 EWOD devices with DC actuation cycles of different polarities: +DC and -DC. No data could be collected for Ta_2O_5 with +DC because electrolysis occurred instantly once voltage was turned on.

Figure 7 shows the contact angle reduction with AC signals of different frequencies applied to Ta_2O_5 EWOD devices. The on- and off-state duration in each cycle is the same as for the DC cases above. Figure 7(a) summarizes

the results on the Ta₂O₅ device with AC frequencies of 50 Hz, 100 Hz, 250 Hz, and 1 kHz. Among the frequencies used in the experiments, the best performance occurs at 50 Hz (which resembles the -DC performance) and diminishes as frequency increases. In contrast, no obvious difference between 50 Hz and 1 kHz was observed on SiO₂ devices as shown in Figure 7(b).

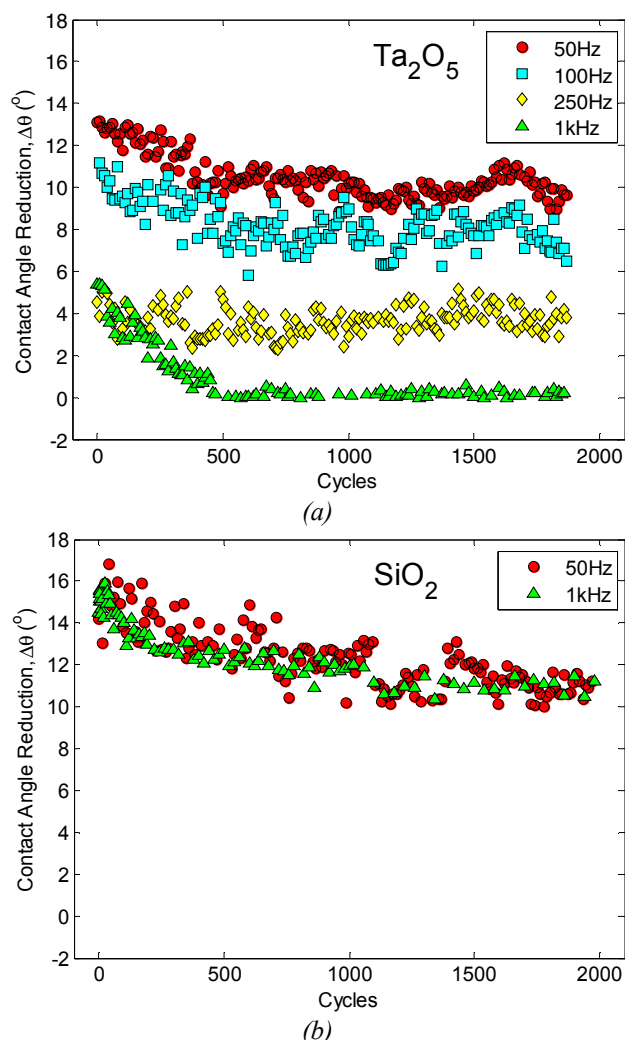


Figure 7: Measured data of electrowetting-induced contact angle reduction on (a) Ta₂O₅ EWOD device and (b) SiO₂ EWOD device with AC signal. Frequencies tested were 50 Hz, 100 Hz, 250 Hz and 1 kHz for Ta₂O₅, and 50 Hz and 1 kHz for SiO₂. The plots show that long-term performance of Ta₂O₅ device worsens with higher AC frequencies, while that of SiO₂ device is not affected much by the AC frequency.

CONCLUSION

EWOD devices made with a Ta₂O₅ dielectric layer exhibit severe polarity and frequency dependencies not observed with common dielectric materials such as SiO₂. Specifically, long-term performance of Ta₂O₅ EWOD devices is acceptable only when the droplet is negatively biased and the actuation frequency is low (< 50 Hz for the current system).

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